

Synthesis and Transformation of Methyl 2-(6-Hydroxy-2-phenyl-pyrimidin-4-yl)acetate: Simple Preparation of Pyrimidines with Heterocyclic Substituents

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A simple and efficient synthesis of four new substituted pyrimidines, compounds **9a–d**, from the title compound **3** is described. Conversion of **3** to methyl (*E*)-3-(dimethylamino)-2-(6-methoxy-2-phenyl-pyrimidin-4-yl)prop-2-enoate (**4**), followed by condensation with various dinucleophiles according to the ‘enaminone methodology’, afforded the target compounds **9** in medium-to-good yields.

Introduction. – Pyrimidines with aromatic or non-aromatic heterocyclic groups are an important class of compounds, which have found application as ligands for metal-ion coordination [1], as components for molecular recognition [2], as therapeutic agents [3], and as fluorophores and electron transporters in light-emitting organic devices [4]. Recently, compounds based on a pyridinylpyrimidine core structure were identified as inhibitors of both human methionine aminopeptidases MetAP1 and MetAP2 [5].

There are many methods described in the literature for the preparation of pyrimidines [6]. The synthesis of many aryl- and heteroaryl-substituted pyrimidines has been accomplished by cyclization of acyclic precursors [7]. Palladium (Pd)-catalyzed reactions are a powerful tool for the synthesis of heterocycles, especially for the formation of C–C bonds in aryl- and heteroaryl-substituted heterocycles [8], and readily allow one to introduce diversity to heterocyclic ring systems [9][10]. Most Pd-catalyzed couplings used for carbocyclic systems employ organic bromides, iodides, and triflates as substrates; unfortunately, the corresponding chlorides are unreactive under typical coupling conditions [11]. In some instances, though, halopyrimidines [12] and 5-pyrimidinylboronic acid derivatives [13] have been used in Pd-catalyzed *Suzuki–Miyaura* cross-coupling reactions [14].

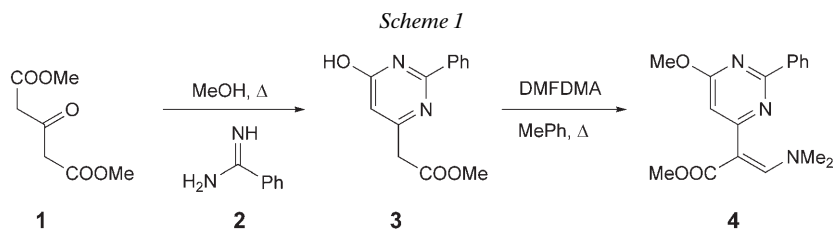
Recently, an expedient parallel synthesis of 2-amino-4-heteroarylpyrimidines *via* a 2-chloropyrimidine intermediate was developed, and a series of potentially biologically active congeners were synthesized in two steps [15].

In connection with our interest in enaminones as building blocks for the preparation of various functionalized heterocyclic systems such as heteroaryl-substituted α -amino- and α -hydroxy acids, fused pyridinones, pyrimidinones, pyranones, and related systems [16], including some naturally occurring alkaloids and their analogues [17], we recently reported the transformation of (*Z*)-3-(benzoylamino)-4-(dimethylamino)-2-oxobut-3-ene by treatment with amidines, guanidines, and guanidine-like compounds into pyrimidine derivatives [18]. We also described the preparation of 6-aminopyrimidine-

4-carboxylic acid derivatives by treatment of methyl 2-(acylamino)-3-cyanopropenoates with methyl- and benzylamine [19], as well as the formation of substituted, fused pyrimidinones from 2-substituted 3-(dimethylamino)propenoates and heterocyclic α -amino compounds in hot acetic acid. Under these conditions, the cyclization of intermediates, 2-substituted 3-heteroarylaminopropenoates, afforded some fused pyrimidines [20]. Finally, a simple and efficient synthesis of novel 2-heteroaryl-substituted 1*H*-indole-2-carboxylates and γ -carbolines from 2-(2-methoxy-2-oxoethyl)-1-methyl-1*H*-indole-3-carboxylate by the ‘enaminone methodology’ was reported recently [21].

In this communication, we report a simple, Pd-free synthesis of pyrimidinylacetic acid derivatives and their transformation into heteroaryl-substituted pyrimidines.

Results and Discussion. – As shown in *Scheme 1*, dimethyl 3-oxopentanedioate (**1**) was reacted with benzamidine (**2**) to methyl 2-(6-hydroxy-2-phenylpyrimidin-4-yl)acetate (**3**). In the subsequent reaction with *N,N*-dimethylformamide dimethylacetal (DMFDMA) in toluene at reflux, the reactive methylene group of **3** was converted to the corresponding *N,N*-dimethylaminomethylidene derivative, under simultaneous methylation of the OH group, to afford the (*E*)-configured propenoate **4**.



Theoretically, the reaction of **3** with DMFDMA could afford three different congeners, depending on the site of methylation. It is known from the literature [22] that, in most cases, methylation of cyclic lactams with this reagent takes place at the ring N-atom. Accordingly, one would expect the formation of the N(1)- or N(3)-methylated compounds, or a mixture thereof. However, the ¹H-NMR chemical shift (δ (H) 3.61) of the newly introduced Me group strongly indicated that, as a matter of fact, *O*-methylation had taken place, giving rise to the formation of **4**.

The configuration of the C=C bond in **4** could not be determined on the basis of the ³*J*(¹³C,¹H) coupling between the ester C=O group and the olefinic H-atom, due to broadening of the methylidene signal. However, an X-ray crystal-structure analysis of **4** clearly confirmed the (*E*)-configuration, as shown in the *Figure*.

Some transformations of **4** with different nucleophiles were studied next (*Scheme 2*). In the reaction with aqueous MeOH in the presence of HCl, hydrolysis took place to give the corresponding enol **5**. In contrast, N-nucleophiles such as methyl glycinate (**6a**), aniline (**6b**), 4-methylaniline (**6c**), and 4-nitroaniline (**6d**), when reacted with **4** in MeOH in the presence of HCl at room temperature, afforded the corresponding substitution products **7a–d**, respectively, in different yields (*Table 1*).

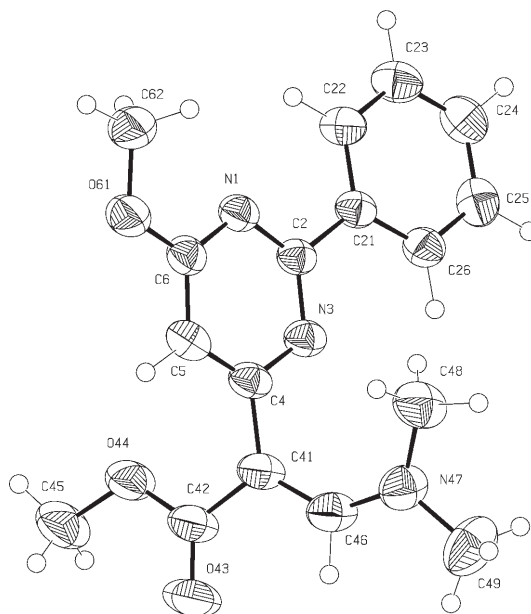


Figure. X-Ray crystal structure of **4**. ORTEP Plot at the 50%-probability level; H-atoms are drawn as ellipsoids of arbitrary radii.

In solution, compounds **7** exist as mixtures of (*E*)- and (*Z*)-isomers. In all cases, the (*E*)-isomer was found to predominate (Table 1). The configuration of (*E*)-**7a** was determined on the basis of the $^3J(\text{C,H})$ coupling constant (4.5 Hz) between the methyldene H-atom and the ester C=O group. Further, the $^1\text{H-NMR}$ spectra of compounds **7** indicated antiperiplanar orientations of the H-atoms in the NH–CH structural elements of the (*E*)- and (*Z*)-isomers, with $J(\text{NH,CH})$ values of 12.6–13.4 Hz (*E*) and 13.0–14.5 Hz (*Z*).

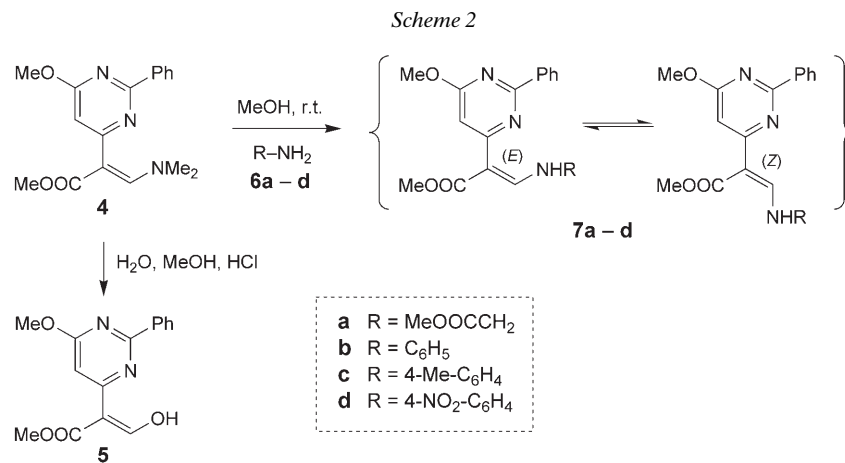
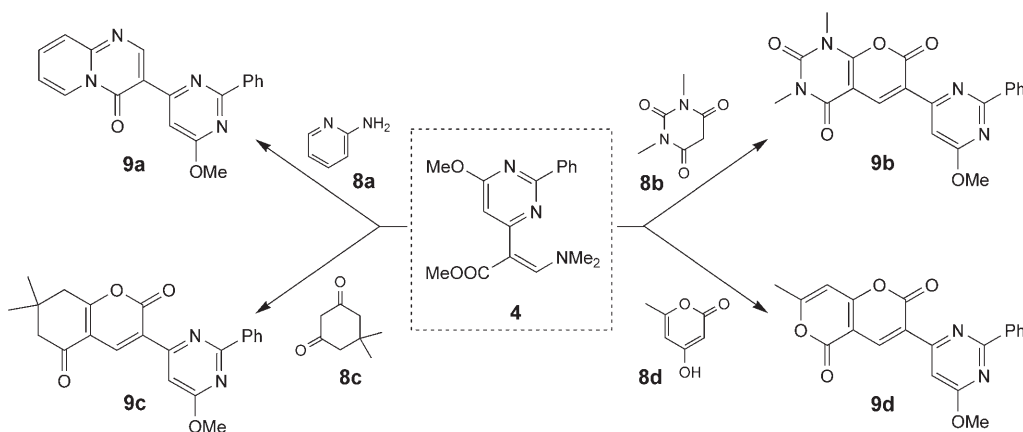


Table 1. Selected Data of Compounds **7**

| No. | Yield [%] | (E/Z) | (E)-Isomer | | | (Z)-Isomer | | |
|-----------|-----------|-------|----------------------------|--------------------------|-------|----------------------------|--------------------------|-------|
| | | | $\delta(\text{H})$ C=CH | $\delta(\text{H})$ NH | 3J | $\delta(\text{H})$ C=CH | $\delta(\text{H})$ NH | 3J |
| 7a | 81 | 66:34 | 8.17 | 10.82 | 13.4 | 8.63 | 9.01 | 14.5 |
| 7b | 48 | 78:22 | 8.62 | 12.89 | 12.9 | 9.02 | 10.56 | 14.0 |
| 7c | 63 | 87:13 | 8.58 | 12.88 | 12.9 | 9.01 | 10.52 | 14.1 |
| 7d | 22 | 72:28 | 8.61 | 12.70 | 12.6 | 8.92 | 10.66 | 13.0 |

Finally, reaction of **4** with the dinucleophiles 2-aminopyridine (**8a**), 1,3-dimethylbarbituric acid (**8b**), 5,5-dimethylcyclohexane-1,3-dione (**8c**), or 4-hydroxy-6-methyl-2H-pyran-2-one (**8d**) was carried out in AcOH at reflux, which readily afforded, through substitution and intramolecular cyclization, the target compounds **9a–d** in yields of 30, 67, 56, and 73%, respectively (Scheme 3). All compounds were fully characterized (see *Exper. Part*).

Scheme 3



Financial support by the *Slovenian Research Agency* (grants P1-0179 and J1-6689-0103-04) as well as by the pharmaceutical companies *LEK-SANDOZ* (Ljubljana) and *KRKA* (Novo Mesto) is gratefully acknowledged. We also thank the Ministry of Science and Technology, Republic of Slovenia, for grants X-2000 and PS-511-103, which allowed the Laboratory of Inorganic Chemistry (Faculty of Chemistry and Chemical Technology, University of Ljubljana, Slovenia) to purchase a *Nonius Kappa CCD* diffractometer, on which the diffraction data of **4** were collected.

Experimental Part

General. All chemicals and reagents were purchased from *Fluka* or *Aldrich*, and used as received. Melting points (m.p.) were determined on a *Kofler* hot-stage micro-melting-point apparatus. IR Spectra were recorded on a *Perkin-Elmer-BX* FT-IR spectrometer; in cm^{-1} . ^1H - and ^{13}C -NMR Spectra were recorded on a *Bruker Avance DPX-300* apparatus at 300 (^1H) and 75.5 MHz (^{13}C); δ in ppm rel. to Me_4Si , J in Hz. Elemental analyses were performed on a *Perkin-Elmer CHN Analyzer 2400-II*.

Methyl 2-(6-Hydroxy-2-phenylpyrimidin-4-yl)acetate (3). To a soln. of MeONa (500 mg Na dissolved in 20 ml MeOH), benzamidine hydrochloride (**2**) (3.132 g, 20 mmol) was added, and the mixture was stirred at r.t. for 10 min. Then, dimethyl 3-oxopentanedioate (**1**; 3 ml, 20 mmol) was added, and the mixture was heated at reflux for 7 h. After cooling, the product precipitated, and was filtered off. Yield: 3.661 g (75%). M.p. 172–173°. IR: 3070, 2960, 1740, 1660, 1540, 1440, 1330, 1170, 990, 700, 600, 530. ¹H-NMR ((D₆)DMSO): 3.66 (s, MeO); 3.69 (s, CH₂); 6.34 (s, H–C(5)); 7.49–7.60 (m, 3 arom. H); 8.08–8.10 (m, 2 arom. H); 12.59 (br. s, OH). Anal. calc. for C₁₃H₁₂N₂O₃ (244.25): C 63.93, H 4.95, N 11.47; found: C 64.08, H 4.89, N 11.75.

Methyl (2E)-3-(Dimethylamino)-2-(6-methoxy-2-phenylpyrimidin-4-yl)prop-2-enoate (4). To a soln. of **3** (488 mg, 2 mmol) in toluene (5 ml) was added Me₂NCH(OMe)₂ (DMFDMA; 0.86 ml, 6 mmol), and the mixture was heated at reflux for 7 h. After cooling, the solvent was removed *in vacuo*. The residue was taken up in MeOH (ca. 10 ml), and the mixture was cooled to –30°. The resulting precipitate was removed by filtration, and recrystallized from MeOH/H₂O. Yield: 372 mg (60%). M.p. 105–107°. IR: 2950, 2920, 1680, 1590, 1570, 1330, 1220, 1100, 860, 760, 700, 670. ¹H-NMR ((D₆)DMSO): 2.89 (br s, Me₂N); 3.61 (s, MeO); 4.02 (s, MeO); 6.72 (s, H–C(5)); 7.50–7.52 (m, 3 arom. H); 7.82 (br. s, =CH); 8.36–8.40 (m, 2 arom. H). Anal. calc. for C₁₇H₁₉N₃O₃ (313.35): C 65.16, H 6.11, N 13.41; found: C 65.37, H 6.21, N 13.42.

Methyl (2E)-3-Hydroxy-2-(6-methoxy-2-phenylpyrimidin-4-yl)prop-2-enoate (5). To a soln. of **4** (157 mg, 0.5 mmol) in MeOH (2 ml) was added conc. HCl (3 drops), the mixture was left at r.t. for 12 h. The resulting precipitate was removed by filtration and recrystallized from MeOH. Yield: 103 mg (57%). M.p. 134–135°. IR: 3130, 2950, 1690, 1620, 1530, 1480, 1390, 1330, 1270, 1180, 1070, 850, 690, 630. ¹H-NMR ((D₆)DMSO): 3.71 (s, MeO); 4.10 (s, MeO); 7.49 (s, H–C(5)); 7.66–7.76 (m, 3 arom. H); 8.17–8.20 (m, 2 arom. H); 9.44 (s, =CH). Anal. calc. for C₁₅H₁₄N₂O₄ (286.28): C 62.93, H 4.93, N 9.79; found: C 63.11, H 5.04, N 9.65.

General Procedure (GP 1) for the Synthesis of Compounds 7. A soln. of **4** (157 mg, 0.5 mmol) and the appropriate amine **6** (0.5 mmol) in MeOH (2 ml) was left at r.t. for 12 h. The resulting product **7**, which precipitated from the mixture, was filtered off and recrystallized as indicated below.

Methyl (E/Z)-3-[(2-Methoxy-2-oxoethyl)amino]-2-(6-methoxy-2-phenylpyrimidin-4-yl)prop-2-enoate (7a). According to GP 1, with methyl glycinate hydrochloride (**6a**; 63 mg, 0.5 mmol). Yield: 145 mg (81%). (*E/Z*)-Ratio: 66:34. M.p. 128–130° (MeOH). IR: 3490, 3150, 3000, 2950, 1760, 1690, 1640, 1570, 1430, 1330, 1220, 1130, 1043, 850, 780, 690, 570. ¹H-NMR ((D₆)DMSO; (*E*)-isomer): 3.69 (s, MeO); 3.73 (s, MeO); 4.02 (s, MeO); 4.50 (*d*, *J* = 5.9, CH₂); 7.49–7.54 (m, 3 arom. H); 7.57 (s, H–C(5)); 8.17 (*d*, *J* = 13.5, =CH); 8.32–8.35 (m, 2 arom. H); 10.82 (*td*, *J* = 13.4, 5.8, NH). ¹H-NMR ((D₆)DMSO; (*Z*)-isomer): 3.71 (s, MeO); 3.77 (s, MeO); 4.00 (s, MeO); 4.41 (*d*, *J* = 6.1, CH₂); 7.14 (s, H–C(5)); 7.49–7.54 (m, 3 arom. H); 8.44–8.48 (m, 2 arom. H); 8.63 (*d*, *J* = 14.6, =CH); 9.01 (*td*, *J* = 14.5, 6.1, NH). Anal. calc. for C₁₈H₁₉N₃O₅ (357.36): C 60.50, H 5.36, N 11.76; found: C 60.51, H 5.54, N 11.96.

Methyl (E/Z)-2-(6-Methoxy-2-phenylpyrimidin-4-yl)-3-(phenylamino)prop-2-enoate (7b). According to GP 1, with aniline hydrochloride (**6b**; 65 mg, 0.5 mmol). Yield: 87 mg (48%). (*E/Z*)-Ratio: 78:22. M.p. 132–134° (MeOH/DMF). IR: 3110, 1690, 1630, 1560, 1500, 1330, 1260, 1130, 1040, 960, 860, 750, 690, 570, 510. ¹H-NMR ((D₆)DMSO; (*E*)-isomer): 3.76 (s, MeO); 4.05 (s, MeO); 7.14–7.20 (m, 1 arom. H); 7.28–7.31 (m, 2 arom. H); 7.42–7.48 (m, 3 arom. H); 7.59 (s, H–C(5)); 8.28–8.32 (m, 2 arom. H); 8.62 (*d*, *J* = 12.9, =CH); 8.32–8.35 (m, 2 arom. H); 12.89 (*d*, *J* = 13.0, NH). ¹H-NMR ((D₆)DMSO; (*Z*)-isomer): 3.85 (s, MeO); 4.03 (s, MeO); 7.14–7.20 (m, 1 arom. H); 7.17 (s, H–C(5)); 7.52–7.54 (m, 3 arom. H); 7.60–7.66 (m, 2 arom. H); 8.41–8.44 (m, 2 arom. H); 9.02 (*d*, *J* = 14.0, =CH); 10.56 (*d*, *J* = 14.0, NH). Anal. calc. for C₂₁H₁₉N₃O₃ (361.39): C 69.79, H 5.30, N 11.63; found: C 69.96, H 5.41, N 11.66.

Methyl (E/Z)-2-(6-Methoxy-2-phenylpyrimidin-4-yl)-3-[(4-methylphenyl)amino]prop-2-enoate (7c). According to GP 1, with 4-methylaniline hydrochloride (**6c**; 72 mg, 0.5 mmol): 119 mg (63%). (*E/Z*)-Ratio: 87:13. M.p. 144–147° (MeOH/DMF). IR: 2950, 1690, 1630, 1570, 1520, 1330, 1260, 1200, 1130, 1050, 860, 810, 700, 510. ¹H-NMR ((D₆)DMSO; (*E*)-isomer): 2.30 (s, Me); 3.75 (s, MeO); 4.04 (s, MeO); 7.16–7.19 (m, 2 arom. H); 7.24–7.27 (m, 2 arom. H); 7.59–7.63 (m, 3 arom. H); 7.59 (s, H–C(5)); 8.26–8.29 (m, 2 arom. H); 8.58 (*d*, *J* = 12.9, =CH); 8.32–8.35 (m, 2 arom. H); 12.88 (*d*, *J* = 12.9, NH). ¹H-NMR ((D₆)DMSO; (*Z*)-isomer): 2.30 (s, Me); 3.84 (s, MeO); 4.02 (s, MeO); 7.26 (s, H–C(5)); 7.28–7.31 (m, 2 arom. H); 7.51–7.53 (m, 2 arom. H); 7.59–7.63 (m, 3 arom. H); 8.40–8.43 (m, 2 arom. H); 9.01

($d, J = 14.1$, =CH); 10.52 ($d, J = 14.1$, NH). Anal. calc. for $C_{22}H_{19}N_3O_3$ (375.42): C 70.38, H 5.64, N 11.19; found: C 70.67, H 5.77, N 11.06.

Methyl (E/Z)-2-(6-Methoxy-2-phenylpyrimidin-4-yl)-3-[(4-nitrophenyl)amino]prop-2-enoate (7d). According to *GP 1*, with 4-nitroaniline hydrochloride (**6d**; 87 mg, 0.5 mmol). Yield: 45 mg (22%). (*E/Z*)-Ratio: 72:28. M.p. 227–230° (MeOH/DMF). IR: 3100, 2940, 1700, 1630, 1590, 1560, 1500, 1330, 1300, 1110, 1040, 870, 840, 750, 700. $^1\text{H-NMR}$ ((D_6) DMSO; (*E*)-isomer): 3.80 (s, MeO); 4.08 (s, MeO); 7.46–7.64 (m , 2 arom. H); 7.62–7.67 (m , 3 arom. H); 7.63 (s, H–C(5)); 8.27–8.34 (m , 4 arom. H); 8.61 ($d, J = 12.5$, =CH); 12.70 ($d, J = 12.6$, NH). $^1\text{H-NMR}$ ((D_6) DMSO; (*Z*)-isomer): 3.88 (s, MeO); 4.05 (s, MeO); 7.17 (s, H–C(5)); 7.46–7.64 (m , 2 arom. H); 7.54–7.56 (m , 2 arom. H); 7.62–7.67 (m , 3 arom. H); 8.44–8.48 (m , 2 arom. H); 8.92 ($d, J = 13.0$, =CH); 10.66 (br. s, NH). Anal. calc. for $C_{21}H_{18}N_4O_5$ (406.39): C 62.06, H 4.46, N 13.79; found: C 62.22, H 4.56, N 13.75.

General Procedure (GP 2) for the Synthesis of Compounds 9. A mixture of **4** (157 mg, 0.5 mmol) and the appropriate dinucleophile **8** (0.5 mmol) in AcOH (2 ml) was heated at reflux for 1.0–4.5 h. After cooling, the product precipitated, and was filtered off and recrystallized as indicated below.

3-(6-Methoxy-2-phenylpyrimidin-4-yl)-4H-pyrido[1,2-a]pyrimidin-4-one (9a). According to *GP 2*, with 2-aminopyridine (**8a**; 47 mg, 0.5 mmol) for 4.5 h. Yield: 50 mg (30%). M.p. 225–227° (toluene). IR: 3110, 1690, 1630, 1590, 1570, 1480, 1370, 1320, 1290, 1030, 870, 780, 700, 660, 590. $^1\text{H-NMR}$ ((D_6) DMSO): 4.09 (s, MeO); 7.54–7.60 (m , H–C(6), 3 arom. H); 7.88–7.91 (m , H–C(8)); 8.00 (s, H–C(5')); 8.16 ($ddd, J = 8.6, 6.8, 1.6$, H–C(7)); 8.52–8.55 (m , 2 arom. H); 9.24–9.27 (m , H–C(5)); 9.68 (s, H–C(2)). Anal. calc. for $C_{19}H_{14}N_4O_2$ (330.34): C 69.08, H 4.27, N 16.96; found: C 69.03, H 4.25, N 17.00.

6-(6-Methoxy-2-phenylpyrimidin-4-yl)-1,3-dimethyl-2H-pyrano[2,3-d]pyrimidine-2,4,7(1H,3H)-trione (9b). According to *GP 2*, with 1,3-dimethylbarbituric acid (**8b**; 78 mg, 0.5 mmol) for 2 h. Yield: 132 mg (67%). M.p. 277–279° (toluene). IR: 3450, 3130, 1760, 1720, 1680, 1560, 1540, 1440, 1330, 1290, 1160, 1070, 950, 860, 770, 750, 700, 480. $^1\text{H-NMR}$ ((D_6) DMSO): 2.50 (s, MeN; masked by solvent signal); 3.46 (s, MeN); 4.10 (s, MeO); 7.56 (s, H–C(5')); 7.58–7.61 (m , 3 arom. H); 8.44–8.47 (m , 2 arom. H); 9.20 (s, H–C(4)). Anal. calc. for $C_{20}H_{16}N_4O_5$ (392.36): C 61.22, H 4.11, N 14.28; found: C 61.50, H 4.19, N 14.16.

7,8-Dihydro-3-(6-methoxy-2-phenylpyrimidin-4-yl)-7,7-dimethyl-2H-1-benzopyran-2,5(6H)-dione (9c). According to *GP 2*, with 5,5-dimethylcyclohexane-1,3-dione (**8c**; 70 mg, 0.5 mmol) for 2 h. Yield: 106 mg (56%). M.p. 339–343° (toluene). IR: 2960, 1740, 1690, 1650, 1560, 1380, 1200, 970, 870, 700, 590, 520. $^1\text{H-NMR}$ ((D_6) DMSO): 1.10 (s, 2 Me); 1.24 (s, $\text{CH}_2(8)$); 3.28 (s, MeO); 7.30 (br. s, 1 arom. H); 7.55–7.63 (m , H–C(5'), 3 arom. H); 8.16–8.18 (m , 2 arom. H); 8.91 (s, H–C(4)). Anal. calc. for $C_{22}H_{20}N_2O_4$ (376.41): C 70.20, H 5.36, N 7.44; found: 69.92, H 5.02, N 7.74.

3-(6-Methoxy-2-phenylpyrimidin-4-yl)-7-methyl-2H,5H-pyrano[4,3-b]pyran-2,5-dione (9d). According to *GP 2*, with 4-hydroxy-6-methyl-2H-pyran-2-one (**8d**; 63 mg, 0.5 mmol) for 1 h. Yield: 132 mg (73%). M.p. 236–237° (toluene). IR: 3460, 3100, 1740, 1590, 1560, 1540, 1400, 1350, 1210, 1040, 990, 870, 780, 700, 660, 560, 490. $^1\text{H-NMR}$ ((D_6) DMSO): 2.41 (s, Me); 4.10 (s, MeO); 6.80 (s, H–C(5')); 7.58–7.60 (m , 3 arom. H); 7.64 (s, H–C(8)); 8.45–8.48 (m , 2 arom. H); 9.05 (s, H–C(4)). Anal. calc. for $C_{20}H_{14}N_2O_5$ (362.34): C 66.30, H 3.89, N 7.77; found: C 66.54, H 3.95, N 7.62.

*X-Ray Crystal Structure of Methyl (2E)-3-(Dimethylamino)-2-(6-methoxy-2-phenylpyrimidin-4-yl)prop-2-enoate (4)*¹. Single-crystal X-ray diffraction data of **4** were collected at r.t. on a *Nonius Kappa CCD* diffractometer, using the *Nonius Collect* software [23]. *DENZO* and *SCALEPACK* [24] were used for indexing and scaling of the data, and the structure was solved by means of *SIR97* [25]. Refinement was done with the *Xtal3.4* [26] program package. The crystal structure was refined on *F* values using the full-matrix least-squares procedure. The non-H-atoms were refined anisotropically in all cases, and the positions of the H-atoms were geometrically calculated, their positional and isotropic atomic displacement parameters not being refined. Absorption correction was not necessary. The *Regina* [27] weighting scheme was used. Difference *Fourier* maps did not show any significant features. The resulting crystal

¹) The crystallographic data of **4** have been deposited with the *Cambridge Crystallographic Data Centre* as supplementary publication number CCDC-643147. Copies of the data can be obtained, free of charge, at http://www.ccdc.cam.ac.uk/data_request/cif.

Table 2. Crystallographic and Refinement Data of **4**

| | |
|--|---|
| Formula | C ₁₇ H ₁₉ N ₃ O ₃ |
| M _r [Da] | 313.36 |
| Crystal system | monoclinic |
| Space group | <i>P</i> 2 ₁ / <i>n</i> |
| <i>a</i> , <i>b</i> , <i>c</i> [Å] | 8.52710(10), 19.3629(4), 10.4132(2) |
| β [°] | 107.4884(9) |
| <i>V</i> [Å ³] | 1639.85(5) |
| <i>Z</i> | 4 |
| ρ [Mg m ⁻³] | 1.269 |
| μ [mm ⁻¹] | 0.089 |
| Crystal color and dimension [mm] | colorless, 0.3 × 0.3 × 0.2 |
| Temperature [K] | 293(1) |
| Wavelength [Å] | 0.71073 |
| θ _{max} [°] | 27.42 |
| Integr. and indep. refl. | 19203, 3845 |
| Obs. refl. | 2400 |
| R _{int} | 0.042 |
| Threshold criterion | <i>I</i> > 2.0σ(<i>I</i>) |
| Refl. parameters | 208 |
| Final <i>R</i> and <i>R</i> _w | 0.057, 0.045 |
| (Δ/σ) _{max} | 0.0002 |
| Δρ _{max} , Δρ _{min} (e Å ⁻³) | – 0.26, 0.32 |

data and details concerning data collection and refinement are given in Table 2. An ORTEP-III [28] plot of the asymmetric unit of **4** is presented in the Figure.

REFERENCES

- [1] S. Leininger, B. Olenyuk, P. S. Stang, *Chem. Rev.* **2000**, *100*, 853.
- [2] F. H. Beijer, H. Kooijman, A. L. Spek, R. P. Sijbesma, E. J. Meijer, *Angew. Chem., Int. Ed.* **1998**, *37*, 75.
- [3] S. R. Piettre, C. Andre, M.-C. Chanal, J.-B. Duceop, B. Lesur, F. Piriou, P. Raboisson, J.-M. Rondeau, C. Schelcher, P. Zimmerman, A. Ganzhorn, *J. Med. Chem.* **1997**, *40*, 4208.
- [4] K.-T. Wong, T. S. Hung, Y. Lin, C.-C. Wu, G.-H. Lee, S. M. Peng, C. H. Chou, Y. O. Su, *Org. Lett.* **2002**, *4*, 513; G. Hughes, C. Wang, A. S. Batsmanov, M. Fearn, S. Frank, M. R. Bryce, I. F. Perepichka, A. P. Monkman, B. P. Lyons, *Org. Biomol. Chem.* **2003**, *1*, 3069.
- [5] X. Hu, A. Addlagatta, B. W. Matthews, J. O. Liu, *Angew. Chem., Int. Ed.* **2006**, *45*, 3772.
- [6] D. J. Brown, in 'Comprehensive Heterocyclic Chemistry', Eds. A. R. Katritzky, C. W. R. S. R. Evans, A. J. Boulton, A. McKillop, Pergamon Press, Oxford, New York, 1984, Vol. 3, p. 57; U. Undheim T. Bennecke, in 'Comprehensive Heterocyclic Chemistry', Eds. A. R. Katritzky, C. W. R. S. R. Evans, A. J. Boulton, Pergamon Press, Oxford, New York, 1996, Vol. 6, p. 93; D. J. Brown, R. F. Evans, W. B. Cowden, M. D. Fenn, 'The Pyrimidines', J. Wiley & Sons, New York, 1994; S. von Angerer, in 'Houben-Weyl: Science of Synthesis, Method of Molecular Transformations', Ed. Y. Yamamoto, Thieme Verlag, Stuttgart 2006, Vol. 16, p. 379; M. G. Hoffmann, A. Nowak, M. Müller, in 'Houben Weyl: Methods of Organic Chemistry', Thieme Verlag, Stuttgart 1998 Vol. E9b, Part 1, p. 379.
- [7] C. Wang, G.-Y. Jung, A. S. Batsmanov, M. R. Bryce, M. C. Pettry, *J. Mater. Chem.* **2002**, *12*, 173; R. Gompper, H. Mair, K. Polborn, *Synthesis* **1997**, 696.
- [8] J. J. Li, G. W. Gribble, 'Palladium in Heterocyclic Chemistry', Pergamon Press, New York, 2000.

- [9] 'Metal-Catalysed Cross-Coupling Reactions'; Eds. F. Diederich, P. Stang, Wiley-VCH, New York, Weinheim, 1998.
- [10] G. Zeni, R. C. Larock, *Chem. Rev.* **2006**, *106*, 4644; G. Zeni, R. C. Larock, *Chem. Rev.* **2004**, *104*, 2285.
- [11] A. F. Littke, C. G. Fu, *Angew. Chem., Int. Ed.* **2002**, *41*, 4176.
- [12] H. Miyaura, A. Suzuki, *Chem. Rev.* **1995**, *95*, 2457; A. Suzuki, in 'Metal-Catalysed Cross-Coupling Reactions', Eds. F. Diederich, P. Stang, Wiley-VCH, New York, Weinheim, 1998, Chapt. 2.
- [13] D. Peters, A.-B. Hornfeldt, S. Gronowitz, *J. Heterocycl. Chem.* **1990**, *27*, 2165.
- [14] N. Saygili, A. S. Batsanov, M. R. Bryce, *Org. Biomol. Chem.* **2004**, *2*, 852.
- [15] M. G. Bursavich, S. Lombardi, A. G. Gilbert, *Org. Lett.* **2005**, *7*, 4113.
- [16] B. Stanovnik, *J. Heterocycl. Chem.* **1999**, *36*, 1581; B. Stanovnik, J. Svete, *Synlett* **2000**, 1077; B. Stanovnik, J. Svete, *Chem. Rev.* **2004**, *104*, 2433; B. Stanovnik, J. Svete, 'Targets in Heterocyclic Systems', Società Chimica Italiana, Rome, 2000, Vol. 4, p. 105; D. Bevk, R. Jakše, J. Svete, A. Golobič, L. Golič, B. Stanovnik, *Heterocycles* **2003**, *61*, 197; D. Bevk, R. Jakše, A. Golobič, L. Golič, A. Meden, J. Svete, B. Stanovnik, *Heterocycles* **2004**, *63*, 609; D. Bevk, L. Golič, A. Golobič, J. Svete, B. Stanovnik, *Heterocycles* **2005**, *66*, 207.
- [17] L. Selič, R. Jakše, K. Lampič, L. Golič, S. Golič Grdadolnik, B. Stanovnik, *Helv. Chim. Acta* **2000**, *83*, 2802; L. Selič, B. Stanovnik, *Tetrahedron* **2001**, *57*, 3159; R. Jakše, V. Krošelj, S. Rečnik, G. Soršak, J. Svete, B. Stanovnik, S. Golič Grdadolnik, *Z. Naturforsch., B* **2002**, *57*, 453; L. Selič, S. Rečnik, B. Stanovnik, *Heterocycles* **2002**, *58*, 577; R. Jakše, J. Svete, B. Stanovnik, A. Golobič, *Tetrahedron* **2004**, *60*, 4601; Z. Časar, D. Bevk, J. Svete, B. Stanovnik, *Tetrahedron* **2005**, *61*, 7508; J. Waggener, S. Golič Grdadolnik, U. Grošelj, A. Meden, B. Stanovnik, J. Svete, *Tetrahedron: Asymmetry* **2007**, *18*, 464; B. Stanovnik, J. Svete, *Mini-Rev. Org. Chem.* **2005**, *2*, 211, and refs. cit. therein.
- [18] U. Bratušek, A. Meden, J. Svete, B. Stanovnik, *ARKIVOC* **2003**, v, 77.
- [19] L. Pizzioli, B. Ornik, J. Svete, B. Stanovnik, *Helv. Chim. Acta* **1998**, *81*, 231.
- [20] B. Stanovnik, J. Svete, *Chem. Rev.* **2004**, *104*, 2433, and refs. cit. therein.
- [21] D. Bevk, U. Grošelj, A. Meden, J. Svete, B. Stanovnik, *Helv. Chim. Acta* **2006**, *89*, 2774.
- [22] B. Stanovnik, M. Tišler, A. Hribar, G. B. Barlin, D. J. Brown, *Aust. J. Chem.* **1981**, *34*, 1729; B. Stanovnik, A. Štimac, M. Tišler, B. Verček, *J. Heterocycl. Chem.* **1982**, *19*, 577; B. Stanovnik, J. Svete, M. Tišler, *J. Heterocycl. Chem.* **1987**, *24*, 1809; J. Svete, B. Stanovnik, M. Tišler, *J. Heterocycl. Chem.* **1989**, *26*, 145.
- [23] Collect Software, Nonius, BV, Delft, The Netherlands, 1998.
- [24] Z. Otwinowski, W. Minor, *Methods Enzymol.* **1997**, *276*, 307.
- [25] A. Altomare, M. C. Burla, M. Camalli, G. L. Cascarano, C. Giacovazzo, A. Guagliardi, A. G. G. Moliterni, G. Polidori, R. Spagna, *J. Appl. Crystallogr.* **1999**, *32*, 115.
- [26] S. R. Hall, G. S. D. King, J. M. Stewart, 'The Xtal3.4 User's Manual', University of Western Australia, Lamb, Perth, 1995.
- [27] H. Wang, B. E. Robertson, 'Structure and Statistics in Crystallography', Ed. A. J. C. Wilson, Adenine Press, New York, 1985.
- [28] M. N. Burnett, C. K. Johnson, 'ORTEP-III: Oak Ridge Thermal Ellipsoid Plot Program for Crystal Structure Illustrations', Oak Ridge National Laboratory Report ORNL-6895, 1996.

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